

# Effect on Hardness and Composition by the Coating of Ni-TiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> & Ni-Al<sub>2</sub>O<sub>3</sub> on Tungsten Carbide Cutting Tool

<sup>1</sup>Sunil Kumar, <sup>2</sup>Kapil, <sup>3</sup>Avnish Kumar

<sup>1,2,3</sup>Dept. of Mechanical Engineering, Uttarakhand university, Dehradun, Uttarakhand, India

## Abstract

Nano-structured composite layers (NCLs) find their potential applications in industry due to increased hardness, wear resistance and corrosion resistance. Many methods such as Chemical vapor deposition (CVD), Physical Vapor Deposition (PVD), spray coatings, electro deposition etc. have been reported in literature for synthesis of NCLs. Grain structures of electrodeposited layer can be controlled by controlling current, pH of the bath solution, agitation etc.; which improves the micro hardness, wear resistance and corrosion resistance of the composite coatings. Today in industry electrodeposited coatings are gaining more importance due to their low cost, ease and simplicity of operation of composite coatings for tribological applications of cutting tools. Ni is one of the most important hard coating materials and is widely used metal matrix. Generally, composites containing dioxide and nitride (like TiO<sub>2</sub> and TiN) are preferred for high wear resistance; high micro hardness, improved corrosion resistance, and high temperature oxidation resistance as compared to pure metal electroplating. TiO<sub>2</sub> nano-particles have good mechanical properties and can be used as secondary phase to improve the hardness of Ni coatings. In this paper Ni-TiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> & Ni-Al<sub>2</sub>O<sub>3</sub> composite coatings are employed on WC cutting tool via electro-deposition. The composite layers are characterized using scanning electron microscopy (SEM), Vickers micro-hardness tester. The surface morphology of the coated layers shows the deposition of fine grained structures (cauli-flower like) at low currents, which increases the strain produced due to lattice defects and act as a basis for increase in hardness. The co deposition of TiO<sub>2</sub> & Al<sub>2</sub>O<sub>3</sub> with Ni is confirmed by Energy Dispersive Spectro-photograph (EDS).

## Keywords

Electrodeposition, Vickers Micro Hardness Tester

## I. Introduction

Electro deposition is the process of coating a thin layer of one metal on top of a different metal to modify its surface properties. Electro-deposition is one of the challenged processes for improvement of the surface. Specially, it is used for the improvement of mechanical properties such as wear and hardness properties of the coating surface. These have the large projected applications for automotive parts, aerospace, printed, electrical contacts, jewelry, musical instruments, soft metal gaskets, and decorative door, light and bathroom fittings. The Nano-sized particles is used by most the researchers for improving, micro hardness, corrosion resistance. In this electro deposition process Ni-TiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> & Ni-Al<sub>2</sub>O<sub>3</sub> composite coatings are employed on WC cutting tool

In economic point of view electro deposition is an appropriate technique which is used in industries. In electro-deposition process the grain size is major concern, this type of composite coating should ideally be developed at lower temperature range by the process of Electro deposition. Again, electro deposition is simple process for operation and by which uniformly deposited on the specimen surface. Electrodeposition appears to be one of the most suitable method because of low cost, simple operation, ease of

control and high production rate.

## II. Experiment Analysis

Electro-deposition is one of the practical methods to make nanostructure materials. Figure 1 shows a simple setup of electro-deposition system. In this system the electro deposition process is controlled by Potentiostat/ Galvanostat (GAMRY Reference600) instrument. A hot plate is used to control the temperature of solution. A specially designed three electrode electro deposition cell was used for electro deposition of Ni-TiO<sub>2</sub> (Fig. 1) where cutting tool was working electrode, Ag/AgCl as reference electrode and pure Ni act as counter electrode (anode). Hard Nickel bath solution was prepared by adding appropriate amount of chemicals in 150 mL distilled water. The beaker was heated on a hot plate and maintained at 45-50°C with continuous magnetic stirring.



Fig. 1: Setup of Electro-Deposition Process

The temperature was maintained by the use of a hot plate and the electro-deposition was controlled by a DC source. Scanning electron microscopy (SEM) can be used for qualitative surface morphology Analysis and energy dispersive spectroscopy (EDS) is a technique to determine the chemical composition of a material. The electrical current was kept a constant value of 24.4 mA for Ni-TiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub> coating and 23 mA for Ni-Al<sub>2</sub>O<sub>3</sub> coating throughout the experiments and the corresponding voltage varied from approximately 0.7-0.85V for both coatings. A Nickel plate was used as anode whereas the prepared specimens were used as cathode. The pH of the plating solution was maintained by adding sulphuric acid (Increasing pH). The pH was maintained (pH 5.76) in both coatings.

### III. Result analysis

#### A. Micro Hardness Measurement

Micro hardness values measured at different points on the surface shows that the Micro- hardness readings are homogeneous. Table 1 shows the variation of micro hardness values measured on the Ni- TiO<sub>2</sub> Al<sub>2</sub>O<sub>3</sub> coated surface 66.3HRC and for Ni- Al<sub>2</sub>O<sub>3</sub> coated surface 55.8 HRC. From the experiment it is clear that with increase in TiO<sub>2</sub> in Nickel bath concentration the micro hardness value increases. For addition of TiO<sub>2</sub> also similar trends were observed. Surface mechanical property is the outcome of particle co-deposition along with Ni grain size and texture obtained by the deposition process. From both the figure it is clear that Ni- TiO<sub>2</sub> Al<sub>2</sub>O<sub>3</sub> composite coating exhibits more hardness as compare to the Ni- Al<sub>2</sub>O<sub>3</sub> coating.

Table 1: Micro Hardness Measurement

Material	Rockwell Hardness
Titanium dioxide and Alumina oxide	66.3HRC
Alumina oxide	55.8 HRC

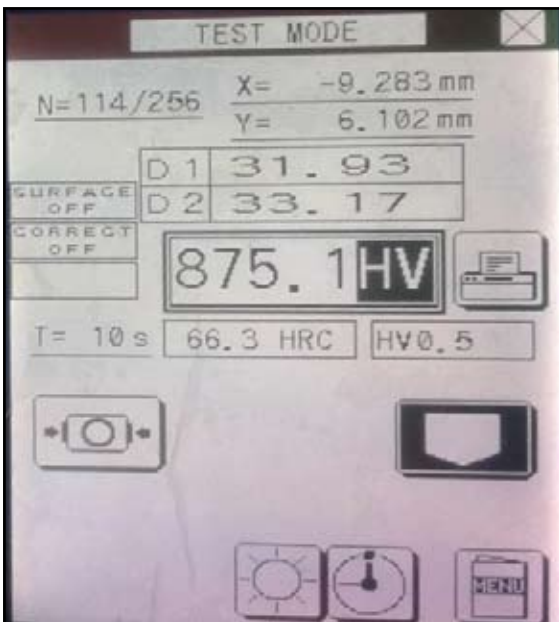


Fig. 2: Micro Hardness of Ni- TiO<sub>2</sub> Al<sub>2</sub>O<sub>3</sub> Coated Specimen

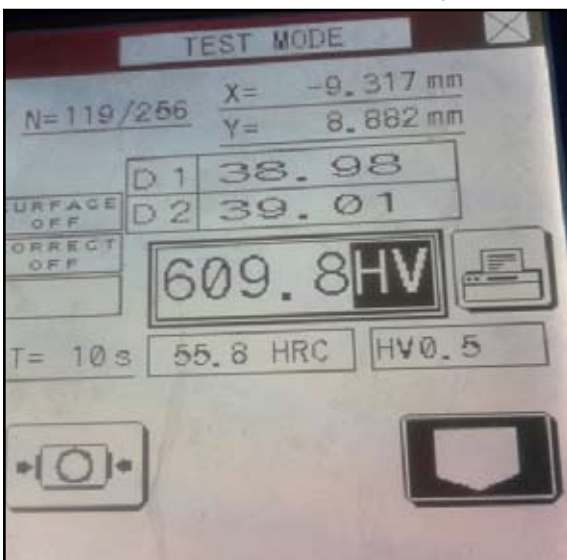


Fig. 3: Micro Hardness of Al<sub>2</sub>O<sub>3</sub> Coated Specimen

#### C. Scanning Electron Microscopy and EDS Result

Microstructure analysis on the coated surface as well as on the cross section was carried out with scanning electron microscope (SEM). Fig. 5, 6 shows SEM micrographs of sample deposited with Ni- TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub> and the Ni-Al<sub>2</sub>O<sub>3</sub>. The particle size of the Ni- TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub> coating is fine as compare to the Ni-Al<sub>2</sub>O<sub>3</sub> coating and its look like a lotus flower and cauli-flower like (dirt –repellent). To determine the chemical composition of the different phases observed in fig. 7, 8 energy dispersive spectroscopy (EDS) study was performed on different region on the coated surface. The deposition of TiO<sub>2</sub> & Al<sub>2</sub>O<sub>3</sub> with Ni is confirmed by Energy Dispersive Spectro-photograph (EDS). In fig. 7, the spectrum confirms the presence of Ti along with Ni and other particles and fig. 8, the spectrum confirms the presence of Al along with Ni.

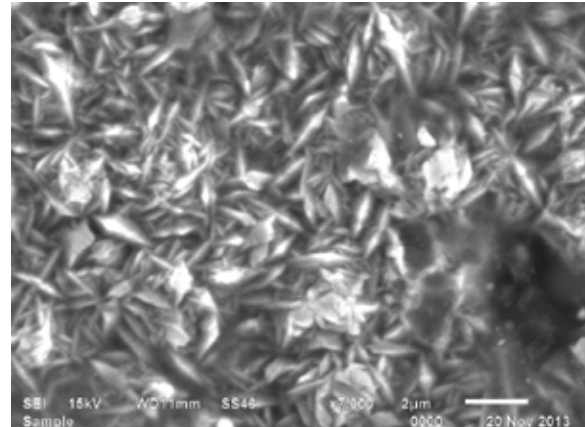


Fig. 4: SEM Micrograph of Tool Bit coated with Ni-TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub>

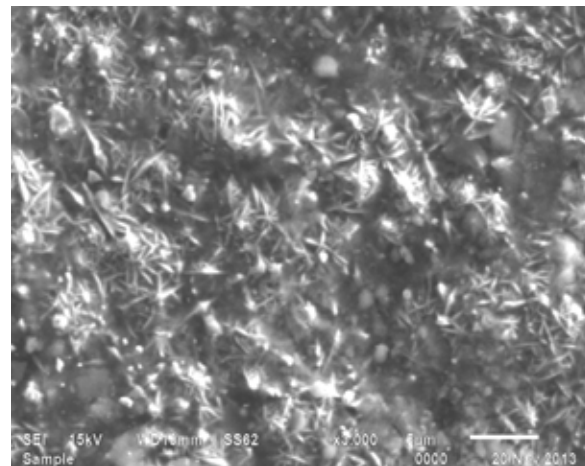


Fig. 5: SEM Micrograph of Tool Bit Coated With Ni-Al<sub>2</sub>O<sub>3</sub>

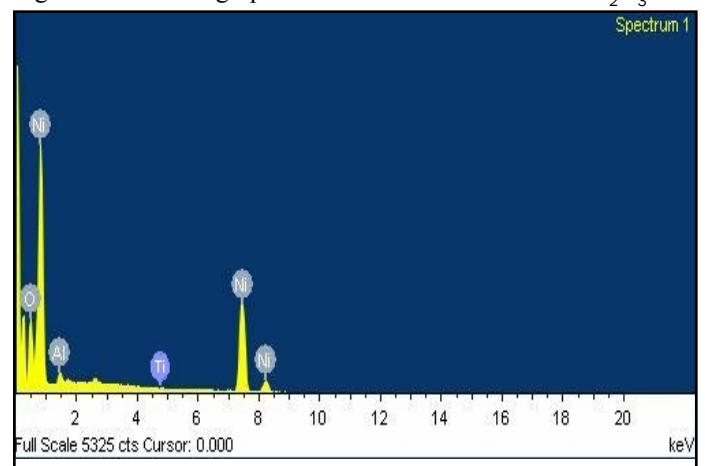


Fig. 6: EDS of Ni-TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub> Coated Specimen

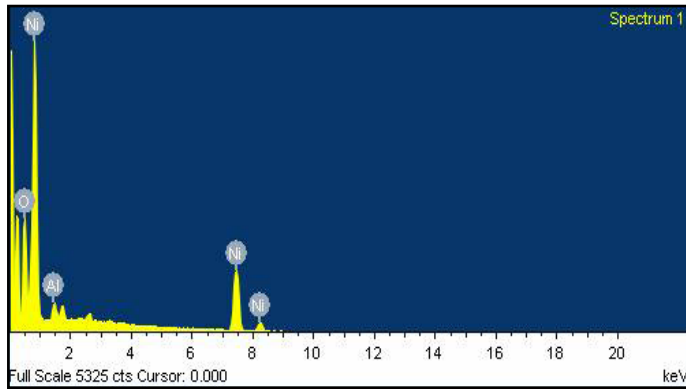


Fig. 7: EDS of Ni-Al<sub>2</sub>O<sub>3</sub> Coated Specimen

#### IV. Conclusion

Ni-TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub> and Ni-Al<sub>2</sub>O<sub>3</sub> coating was done successfully on different Tungsten carbide tool bits by electro-deposition method. The composite layers are characterized using scanning electron microscopy (SEM) and Vickers microhardness tester. Micro hardness measurement of the coating shows that Ni-TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub> coating has high micro hardness value as compared to Alumina oxide coating. Excessive addition of titanium can increased more amount of hardness. The surface morphology of the coated layers shows the deposition of fine grained structures (cauli-flower like) at low currents, which increases the strain produced due to lattice defects and act as a basis for increase in hardness. The deposition of TiO<sub>2</sub> & Al<sub>2</sub>O<sub>3</sub> with Ni is confirmed by Energy Dispersive Spectrophotograph (EDS)

The hardness of the resultant coatings is found to be 66.3HRC for Ni-TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub> coating and 55.8 HRC for Al<sub>2</sub>O<sub>3</sub> coating depending on the particle volume in the Ni matrix. The SEM shows that the particle size of the Ni-TiO<sub>2</sub>Al<sub>2</sub>O<sub>3</sub> coating is fine as compare to the Ni-Al<sub>2</sub>O<sub>3</sub> coating and its look like a lotus flower and cauli-flower (dirt-repellent).

#### Reference

- [1] Agarwala R. C, Agarwala V., "Electroless alloy/composite coatings: A review", Sadhana, Vol. 28, Parts 3 & 4, pp. 475-493, 2003.
- [2] Akhlaghi S., Ivey G.D, "Effect of Processing Parameters on the Electroplating of Au-Sn Solders", Micralyne Inc., Street, Edmonton, Alberta, Canada, pp.1-9, 2003.
- [3] Aytac, M.U., Sert H., "Analysis by Scratch Method of Coatings of AISI5115 and M31 Steels Coated with Al TiN and Cr N Using PVD Method", Leeds-Lyon Symposium on Tribology & tribochemistry, Vol. 40, 2013
- [4] Balaji R., "Preparation and Characterization of Electrodeposits from Methane Sulphonic Acid Bath", Ph.D. Thesis, Industrial Metal Finishing Division Central Electrochemical Research Institute Tamil Nadu, 2006.
- [5] Bestetti Massimiliano, Forno D.A., "Electroless and Electrochemical Deposition of Metallic Coatings on Magnesium Alloys Critical Literature Review", Frank Czerwinski (Ed.), InTech, pp. 154-184, 2011.
- [6] Carlsson T.E., Strand, F., Lindstrom, B., "Statistical model for prediction of tool life as a basis for economical optimization of the cutting process", CIRP Annals, Vol. 41, pp. 79-82, 1992.
- [7] Chauhan K.S, Lakra S.K, "Effect of Substrate Texture on Electroplating", Project Report, N.I.T. Rourkela, 2010.
- [8] Chonglun Fan, Piron D. L., "Study of anomalous nickel-cobalt electro-deposition with different electrolytes and

current densities", pp. 1713-1719, 1996.

- [9] Eltoum M. S. Ali, Baraka A. M., Saber M., Elfatih M. Hassan A., "Electrodeposition And Characterization of Nickel-Titania Nanocomposite Coatings From Gluconate Baths", International Journal of Multidisciplinary Sciences And Engineering, Vol. 2, No. 4, pp. 1-6, 2011.
- [10] Erne Martin, Kolar Daniel, "Thermal Spraying of Oxide Ceramic and Ceramic Metallic Coatings", Applications in Engineering, Prof. Feng Shi (Ed.), InTech, (2012), pp. 167-194, 2012.
- [11] Frankel Gerald S., "Electrochemical Techniques in Corrosion: Status, Limitations, and Needs", Journal of ASTM International, Vol. 5, No. 2, 2008.
- [12] Grzesik W., Zalisz Z. and Nieslony, P. (2002) "Friction and wear testing of multiplayer coatings on carbide substrates for dry machining applications", Surface & Coatings Technology, Vol. 155, pp.37-45.
- [13] Hogmark S, Hedenqvist, P, and Jacobson S. (1997) "Tribological properties of thin hard coatings-demand and evaluation", Surface & Coatings Technology, vol.90, pp, 247- 257.
- [14] Long J.M., Haynes D.A. and Hodgson P.D., (2004) "Characterisation of Galvanneal Coatings on Strip Steel", Materials Forum, Vol. 27, (2004), pp. 62- 67.
- [15] Lou Helen H. and Huang Yinlun, (2006) "Book of Electroplating", Encyclopedia of Chemical Processing copyright by Taylor & Francis, pp.1-10.
- [16] JE J. H., Gyarmati E. and Naoumidis A., (1986) "Scratch Adhesion Test Of Reactively Sputtered Tin Coatings On A Soft Substrate", Thin Solid Films, Vol.136, pp.57-67.



Sunil Kumar received his B.Tech degree in Mechanical from Kurukshetra University Kurukshetra, in 2009, the M.Tech. Degree in CAD/CAM from Punjab Technical University, in 2013. He was a teaching lecturer, in Apex Institute of Engineering and Technology, Karnal 2009 and 2010 respectively. He was worked as an Assistance professor in Maharishi Markandeshwar University Trust Mullana-Ambala from 2010 to

2013. Now he Working as an Assistant professor in Uttaranchal University, Dehradun from 2014 to till date. His research interests include Nano-coating, composite coating on Different cutting tools.



Kapil received his B.Tech degree in Mechanical from U.P.T University Lucknow, in 2010, the M.Tech. Degree from IIT-Bhu, in 2014 respectively. He was worked as an Lecturer in IITM, Meerut from 2010 to 2012. Now he Working as an Assistant professor in Uttaranchal University, Dehradun from 2014 to till date. His research interests include Nano-coating or composite coating on Different cutting tools.



Avnish kumar received his B.Tech degree in Mechanical from U.P.T University Lucknow, in 2010, the M.Tech. Degree from U.P.T University Lucknow, in 2014 respectively. He is working as an Assistant professor in Uttaranchal University, Dehradun from 2014 to till date. His research interests include Heat transfer.